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Modern Chemistry Chapter 3 Review Answers

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone or in molecules containing the

3 Atoms: The Building Blocks of Matter

Particle Mass number 19. Proton 20. Neutron 21. Electron 10 CHAPTER 3 TEST Relative charge Location MODERN CHEMISTRY HRW material copyrighted under notice appearing earlier in this work. Menu Print Name Date Class CHAPTER 3 TEST continued SHORT ANSWER Write the answers to the following questions in the space provided. 22.

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Law of Conservation of mass & proportions, counting atoms related terms. Terms in this set (28) law of conservation of mass. states that mass is neither created nor destroyed during ordinary reactions or physical changes.

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Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

Modern Chemistry Chapter 3 Test B Answers

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter
SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone or in molecules containing the

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Modern Chemistry Chapter 3 1 Review Answers

3:3 mol H₂; or their reciprocals 7. Given the following equation:
 $4\text{NH}_3(\text{g}) + 6\text{NO}(\text{g}) \rightarrow 5\text{N}_2(\text{g}) + 6\text{H}_2\text{O}(\text{g})$
1 mol NO:1 mol H₂O a.
What is the mole ratio of NO to H₂O? 3 mol NO:2 mol NH₃ b.
What is the mole ratio of NO to NH₃? 0.360 mol c. If 0.240 mol of NH₃ react according to the above equation, how many moles of NO will be consumed? 8.

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Modern Chemistry Chapter 5 Section 3 Review Answers

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