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1. Put on your laboratory apron and safety goggles. 2. Place 10 drops of liquid A into each of the six top wells of the 24 well microspot plate. Place 10 drops of liquid B into each of the six wells of the next row. Continue with 10 drops of liquid C in the third row, and 10 drops of liquid D in the bottom row. 3.

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This lab manual accompanies the college chemistry courses and is used to support and assess knowledge acquired in the classroom setting to be applied in the real world. Labs include activities such as measuring the density of a sucrose solution, calculating the percent composition, stoichiometry, determining limiting reactants and more to complement an introductory chemistry course.

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The sequence of experiments in this Laboratory Manual is designed to follow the lecture curriculum. However, instructors will sometimes vary the order of material covered in lecture and thus certain experiments may come before the concepts illustrated are covered in lecture or after the material

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list and pictures of the equipment is provided in Appendix 1 of this laboratory manual. 5. Print and sign the Safety Pledge provided on the next page and turn it in to your instructor. 6 SAFETY PLEDGE Note: Failure to follow safety rules will result in expulsion from this course.

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